Two New Quassinoids, Ailantinols A and B, and Related Compounds from *Ailanthus altissima*

Kengo Kubota,[†] Narihiko Fukamiya,[†] Tomomi Hamada,[†] Masayoshi Okano,^{*,†} Kiyoshi Tagahara,[‡] and Kuo-Hsiung Lee[§]

Department of Interdisciplinary Studies of Natural Environment, Faculty of Integrated Arts and Sciences, Hiroshima University, Higashi-Hiroshima 739, Faculty of Pharmaceutical Sciences, Kobe Pharmaceutical University, Kobe 658, Japan, and Natural Products Laboratory, Division of Medicinal Chemistry and Natural Products, School of Pharmacy, University of North Carolina, Chapel Hill, North Carolina 27599

Received July 7, 1995[®]

Two new quassinoids, ailantinols A (1) and B (2), and related compounds were isolated from *Ailanthus altissima*, and their structures were elucidated from spectral evidence.

Many guassinoids with various biological activities such as antitumor, antimalarial, antifeedant, insecticidal, anti-inflammatory, amoebicidal, and herbicidal effects have been isolated from plants in the Simaroubaceae family.¹ Our research has focused on the isolation of biologically active quassinoids from plants such as Brucea antidysenterica,2-8 Picrasma ailanthoides,9-12 and Brucea javanica.^{13,14} As part of these studies, we investigated the isolation of quassinoids from Ailanthus altissima Swingle (= Ailanthus glandulosa Desf.) (Simaroubaceae). In this paper, we report the isolation and structural elucidation of two new guassinoids, ailantinols A (1) and B (2), from A. altissima. The known quassinoids, shinjudilactone (**3**),^{15,16} ailanthone (4),^{17,18} shinjulactone A,¹⁹ amaloride,^{20,21} amaloride 11acetate,²⁰ shinjulactone K,²² $\Delta^{13(18)}$ -dehydroglaucarubinone,²³ $\Delta^{13(18)}$ -dehydroglaucarubolone,^{24,25} shinjulactone B,^{26,27} and shinjulactone C^{15,23} also were isolated from this plant; all of these compounds have been reported previously as constituents of A. altissima. Compounds 3 and 4 were useful in the structural elucidation of the new compounds 1 and 2.

Compound **1** was isolated as colorless needles. Its IR spectrum showed the presence of hydroxy (3450 cm⁻¹), γ -lactone (1800 cm⁻¹), δ -lactone (1750 cm⁻¹), ester (1739 cm⁻¹), and α,β -unsaturated carbonyl (1670 cm⁻¹) groups. Its UV spectrum showed an absorption maximum at 235 nm due to a conjugated enone system. Its molecular formula was established as C₂₁H₂₆O₈ by HREIMS (*m*/*z* 406.1597). The ¹H- and ¹³C-NMR spectral data of compound **1** did not coincide with those of any known quassinoid, and we have assigned the name ailantinol A to this new quassinoid.

The ¹H-NMR spectrum (Table 1) of **1** was similar to that of **3**, except for the positions of the H-9, H-13, H-14, and OMe signals (δ 3.77, 3.44, 2.82, and 3.60). The former three signals in **1** were shifted downfield by 1.05, 1.26, and 0.54 ppm from the corresponding signals in **3**.

The ¹³C-NMR spectrum (Table 2) of **1** also was similar to that of **3** except for the positions of some signals in



quaternary carbon.

Ĥ

3

and the H-15 α and H-15 β methylene signals (δ 3.01 and 2.68) with the C-13 and C-14 signals (δ 36.9 and 40.9). A mass fragment ion peak of m/z 151 due to a tropylium ion provided further confirmation of the structure of ring A.

the C-ring and of the OMe signal (δ 52.2). The C-13,

C-14, and C-20 signals (δ 36.9, 40.9, and 69.4) in **1**

appeared at higher field than those in **3** by 8.7, 12.7,

and 6.8 ppm. The C-11 signal (δ 175.6) in **1** shifted to

lower field by 96.9 ppm compared with that (δ 78.7) in

3. DEPT spectra of **1** and **3** showed that the former

signal is due to a carbonyl carbon and the latter to a

The partial structures of rings A, B, and D in **1** were

OCH₂

2

Ĥ

4

HC

OH

The structure of the cleaved C-ring in **1** was elucidated from the following evidence. In the ¹H–¹H COSY NMR spectrum, the 13-Me signal (δ 1.31) coupled with the H-13 resonance (δ 3.44), which in turn coupled with the H-14 signal (δ 2.82). The H-14 resonance coupled with the signals (δ 2.68 and 3.01) of the H-15 methylene. In the HMBC spectrum, the ester carbonyl signal (δ



^{*} To whom correspondence should be addressed.

[†] Department of Interdisciplinary Studies of Natural Environment, Hiroshima University.

[‡] Faculty of Pharmaceutical Sciences, Kobe Pharmaceutical University. [§] Natural Products Laboratory, University of North Carolina, Chapel

Hill. Abstract published in Advance ACE Abstracts June 15, 1006

Table 1.	¹ H-NMR	Spectra ^a o	f Com	pounds	1 - 4
----------	--------------------	------------------------	-------	--------	-------

	compound				
proton	1 ^b	2^{b}	3 ^c	4^d	
H-1	4.75 s	4.42 s	4.23 s	4.54 s	
H-3	6.14 br s	6.12 br s	6.10 br s	6.13 br s	
H-5	3.01 m	3.03 br d (13)	3.22 br d (13)	3.07 br d (13)	
Η-6α	2.35 ddd (15.5, 5, 2.5)	2.19 ddd (15, 3.5, 3)	2.31 ddd (15, 3, 2.5)	2.22 br d (14)	
H-6 β	2.10 ddd (15.5, 13.5, 2.5)	2.07 ddd (15, 3, 3)	2.08 ddd (15, 13, 2.5)	2.05, br dd (14, 13)	
H-7	4.91 dd (2.5, 2.5)	4.62 dd (3.5, 3)	4.81 t (2.5)	4.64 br s	
H-9	3.77 s	3.37 s	2.72 s	3.53 s	
H-12		4.16 br d (5)		4.46 s	
H-13	3.44 m		2.18 guin (7)		
H-14	2.82 m	2.55 dd (14.5, 5.5)	2.28 ddd (10.5, 7, 1)	2.83 dd (13, 5)	
Η-15α	3.01 m	3.29 dd (18.5, 14.5)	3.13 dd (16, 10.5)	3.69 dd (18, 13)	
H-15β	2.68 dd (17, 5.5)	3.09 dd (18.5, 5.5)	2.70 dd (16, 1)	2.90 dd (18, 5)	
Η-20α	4.69 d (10.5)	4.78 d (8)	4.79 d (12)	4.11 d (8)	
H-20β	4.30 d (10.5)	4.14 d (8)	4.31 d (12)	3.66 d (8)	
H-21	× ,			5.19 br s	
				5.28 br s	
4-Me	1.77 s	1.76 br s	1.80 br s	1.78 br s	
10-Me	1.19 s	1.61 s	1.24 s	1.52 s	
13-Me	1.31 d (6.5)	1.67 s	1.24 d (7)		
13-COOMe	3.60 s				
12-OH		7.60 d (5)		е	

^{*a*} Values are in ppm. The coupling constants (*J* values) in parentheses are in Hz. ^{*b*} 500 MHz in C₅D₅N. ^{*c*} 400 MHz in C₅D₅N. ^{*d*} 400 MHz in C₅D₅N

Table 2. ¹³C NMR Spectra^a of Compounds 1-4

	compound				
carbon	1 ^b	2^{b}	3 ^c	4^d	
C-1	84.7 (CH)	84.6 (CH)	83.8 (CH)	84.4 (CH)	
C-2	197.5 (C=O)	197.6 (C=O)	196.9 (C=O)	197.3 (C=O)	
C-3	126.9 (CH)	126.3 (CH)	126.3 (CH)	126.2 (CH)	
C-4	160.3 (C)	162.3 (C)	162.0 (C)	162.1 (C)	
C-5	40.6 (CH)	42.6 (CH)	42.2 (CH)	44.8 (CH)	
C-6	26.0 (CH ₂)	26.1 (CH ₂)	27.0 (CH ₂)	26.2 (CH ₂)	
C-7	75.7 (CH)	78.2 (CH)	73.9 (CH)	78.5 (CH)	
C-8	45.8 (C)	46.5 (C)	48.4 (C)	45.7 (C)	
C-9	54.3 (CH)	44.8 (CH)	55.0 (CH)	48.0 (CH)	
C-10	45.0 (C)	45.3 (C)	43.0 (C)	45.5 (C)	
C-11	175.6 (C=O)	110.7 (C)	78.7 (C)	110.3 (C)	
C-12	172.8 (C=O)	83.0 (C=O)	173.5 (C=O)	80.6 (C=O)	
C-13	36.9 (CH)	74.2 (C)	45.6 (CH)	147.4 (C)	
C-14	40.9 (CH)	49.0 (CH)	53.6 (CH)	42.5 (CH)	
C-15	31.4 (CH ₂)	31.8 (CH ₂)	32.9 (CH ₂)	35.3 (CH ₂)	
C-16	172.3 (C=O)	170.2 (C=O)	170.6 (C=O)	169.4 (C=O)	
C-18	22.2 (Me)	22.4 (Me)	22.1 (Me)	22.5 (Me)	
C-19	10.7 (Me)	10.7 (Me)	10.6 (Me)	10.3 (Me)	
C-20	69.4 (CH ₂)	71.0 (CH ₂)	76.2 (CH ₂)	72.3 (CH ₂)	
C-21	10.7 (Me)	26.2 (Me)	13.8 (Me)	118.2 (CH ₂)	
13-COOMe	52.2 (Me)				

^a Values are in ppm (C₅D₅N). ^b 125 MHz. ^c 22.5 MHz. ^d 25 MHz.



Figure 1. $^{13}C^{-1}H$ long-range correlations in the HMBC spectrum of 1.

172.8) showed a ${}^{13}C^{-1}H$ long-range correlation with the proton signals (δ 1.31 and 3.60) of the 13-Me and the OMe. Furthermore, the C-13 signal (δ 36.9) correlated with the H-15 β and 13-Me signals (δ 2.68 and 1.31).

The structure of the γ -lactone moiety in **1** was elucidated from the following evidence. In the HMBC NMR spectrum of **1**, the carbon signal (δ 175.6, C-11)



Figure 2. NOE correlations of 1.

of the γ -lactone C=O showed ${}^{13}\text{C}{-}^{1}\text{H}$ long-range correlation with the H-9 and H-20 β signals (δ 3.77 and 4.30). Furthermore, the H-20 methylene signals (δ 4.30 and 4.69) correlated with the C-8 and C-9 signals (δ 45.8 and 54.3). These data indicated that the γ -lactone moiety bridges C-8 and C-9.

The relative stereochemistry of **1** was confirmed from its NOESY NMR spectrum. The NOE correlations are shown by arrows in Figure 2 and support the proposed structure of **1**.

Compound **2** was obtained as colorless needles. Its IR spectrum showed the presence of hydroxy (3400 cm⁻¹) and δ -lactone (1720 cm⁻¹) groups. Its UV spectrum exhibited maximum absorption at 240 nm due to a conjugated enone system. HREIMS (m/z 394.1624) revealed a molecular formula of C₂₀H₂₆O₈, which is 18 units higher than that of **4** (C₂₀H₂₄O₇). The spectral data of compound **2** also did not correspond to those of any known quassinoid, and we have assigned the name ailantinol B to this new quassinoid.

The ¹H-NMR signals of **2** were similar to those of **4**; however, a methyl signal (δ 1.67) in **2** replaced the H-21 exomethylene signals (δ 5.19 and 5.28) in **4**. Additionally, the H-12 and H-14 signals (δ 4.16 and 2.55) of **2** were shifted upfield by 0.30 and 0.28 ppm, respectively, compared with those of **4**. These findings suggested the presence of a hydroxy group at C-13, which is consistent with the mass spectral data.

The 13 C-NMR signals observed for C-1 to C-12 and C-15 to C-20 in **2** were nearly identical with those of **4**.



Figure 3. ${}^{13}C^{-1}H$ long-range correlations in the HMBC spectrum of **2**.



Figure 4. NOE correlations of 2.

However, in the DEPT NMR spectrum of **2**, C-13 is a quaternary carbon and C-21 is a methyl carbon, whereas these carbons in **4** are both olefinic. The chemical shift value (δ 74.2) of C-13 in **2** is reasonable for a carbon attached to an oxygen atom. From these data, the structure of **2** was proposed as that shown. The ¹³C-¹H long-range correlations found in the HMBC spectrum of **2** are depicted by arrows in Figure 3 and are consistent with the proposed structure.

The relative stereochemistry of **2** was confirmed by its NOESY NMR spectrum; the NOE correlations are shown by arrows in Figure 4.

Experimental Section

General Experimental Procedures. Melting points were determined on an MRK air-bath-type melting point apparatus and are uncorrected. Specific rotations were obtained on a JASCO DIP-370 digital polarimeter (l =0.5 dm). IR and UV spectra were recorded on a JASCO IR-810 spectrometer and Hitachi 320-S spectrometer, respectively. ¹H- and ¹³C-NMR spectra were determined on a Varian VXR-500, a JASCO GSX-500, or a JEOL Alpha-400 spectrometer in C₅D₅N using TMS as an internal standard. MS were recorded on a Hitachi M-80 instrument. Si gel (Merck, type 60, 70–230 mesh) and Sephadex LH-20 (Pharmacia) were used for column chromatography. Precoated Si gel plates (Merck, 60F₂₅₄) of 0.25-mm thickness were used for analytical TLC, and plates of 1-mm and 2-mm thickness were used for preparative TLC. Components were detected using a UV lamp (254 and 365 nm). Analytical HPLC was performed on a Tosoh liquid chromatograph equipped with a UV detector (254 nm) and a reversed-phase column (TSK-gel ODS-80T_s) using a MeOH/H₂O mixture as solvent. Preparative HPLC was carried out on Tosoh or Gilson liquid chromatographs equiped with a reversed-phase column (Dynamax-60A and/or Lichrosorb RP-18) using the same solvents as employed for analytical HPLC.

Plant Material. In 1982, five trees of *A. altissima* were planted in the Higashisenda campus of Hiroshima

University. In 1990, four plants were harvested, and the stem bark was collected for extraction. A voucher specimen was deposited at the above campus.

Extraction and Isolation. The stem bark of A. altissima (fresh material, 33 kg) was cut into small pieces and soaked in MeOH (76 L) for 1 year at room temperature. Evaporation of the solvent gave a MeOH extract (2.5 kg), which was partitioned between MeOH/ H_2O (2:1) and *n*-hexane. The MeOH/ H_2O layer was then extracted with CHCl₃ to afford a CHCl₃ extract (221 g). Si gel column chromatography of this extract (221 g) with EtOAc/Et₂O (1:1, v/v) (13 L) gave 95 fractions and with MeOH (13 L) gave 27 fractions. Each fraction was checked by analytical TLC and HPLC; combination gave 19 subfractions: 1 (5.94 g), 2 (6.67 g), 3 (4.89 g), 4 (5.53 g), 5 (6.42 g), 6 (6.47 g), 7 (4.35 g), 8 (4.99 g), 9 (4.55 g), 10 (4.02 g), 11 (4.85 g), 12 (3.29 g), 13 (5.83 g), 14 (6.01 g), 15 (15.2 g, MeOH), 16 (16.1 g, MeOH), 17 (15.0 g, MeOH), 18 (19.6 g, MeOH), and 19 (18.7 g, MeOH).

Subfraction 12 (3.29 g) was subjected to preparative TLC (EtOAc/Et₂O, 1:1) to give six fractions: 12-1 (0.28 g), -2 (1.77 g), -3 (1.09 g), -4 (0.35 g), -5 (0.31 g), and -6 (0.36 g). Preparative HPLC (MeOH/H₂O, 2:8) of fraction 12-2 (1.77 g) gave nine additional fractions: 12-2-1 (301 mg), -2 (172 mg), -3 (95 mg), -4 (143 mg), -5 (36 mg), -6 (93 mg), -7 (198 mg), -8 (95 mg), and -9 (55 mg). Repeated preparative HPLC (MeOH/H₂O, 3:7) of fraction 12-2-8 (95 mg) afforded the new quassinoid, ailantinol A (1, 15 mg, 0.000 045%), as colorless needles. This compound was also isolated from subfractions 10 and 11; the total amount was 31.0 mg (0.000 95%). Fraction 12-2-6 (93 mg) was subjected to repeated preparative HPLC (MeOH/H₂O, 2:8 and 15:85) to afford ailanthone (4, 32 mg, 0.000 098%). Crystals (18.8 mg) from a MeOH solution of fraction 12-2-4 (143 mg) were purified by preparative HPLC (MeOH/H₂O, 2:8) to afford shinjudilactone (**3**, 16.3 mg, 0.000 049%).

Dissolving subfraction 13 (5.83 g) in MeOH afforded crystals, which were recrystallized from MeOH to afford shinjulactone B (441 mg, 0.0013%). The mother liquor was subjected to column chromatography on Sephadex LH-20 to give four fractions: 13-1 (286 mg), -2 (1.45 g), -3 (2.22 g), and -4 (1.08 g). Preparative HPLC (MeOH/ H_2O , 2:8) of fraction 13-2 (1.45 g) gave seven fractions: 13-2-1 (163 mg), -2 (188 mg), -3 (46 mg), -4 (48 mg), -5 (122 mg), -6 (82 mg), and -7 (636 mg). Fraction 13-2-2 (188 mg) then was subjected to preparative TLC (CHCl₃/ MeOH/H₂O) to give seven fractions: 13-2-2-1 (32 mg), -2 (62 mg), -3 (35 mg), -4 (9 mg), -5 (45 mg), -6 (12 mg), and -7 (8 mg). Fraction 13-2-2-3 (35 mg) was purified by HPLC (MeOH/H₂O, 15:85) to afford the new quassinoid, ailantinol B (2, 7.2 mg, 0.000 021%), as colorless needles. Similar treatment of fraction 13-2-2-4 (9 mg) afforded $\Delta^{13(18)}$ -dehydroglaucarubolone (7.6 mg, 0.000 023%). Fraction 13-3 (2.22 g) was subjected to preparative HPLC to give eight fractions. Crystals obtained from a MeOH solution of fraction 13-3-6 (318 mg) were recrystallized from MeOH to afford shinjulactone B (210 mg, 0.000 63%); the total amount of shinjulactone B was 651 mg (0.0019%).

Ailantinol A (1): colorless needles; mp 193–195 °C; [α]³⁰_D +77° (*c* 0.03, MeOH); UV (EtOH) λ_{max} 235 (ϵ 10 200) nm; IR (KBr) ν_{max} 3450 (OH), 1800 (γ -lactone C=O), 1750 and 1730 (δ -lactone and ester C=O), and 1670 (α,β-unsaturated C=O) cm⁻¹; EIMS m/z [M]⁺ 406 (85%); HREIMS m/z [M]⁺ 406.1611, calcd for C₂₁H₂₆O₈ 406.1597; ¹H- and ¹³C-NMR data, see Tables 1 and 2.

Ailantinol B (2): colorless needles; mp 149–152 °C; [α]²⁶_D +71° (*c* 0.02, MeOH), UV (MeOH) λ_{max} 240 (ϵ 8700) nm; IR (KBr) ν_{max} 3400 (OH), 1720 (δ -lactone C=O), and 1670 (α,β -unsaturated C=O) cm⁻¹; HREIMS m/z [M]⁺ 394.1628, calcd for C₂₀H₂₆O₈ 394.1628; ¹H- and ¹³C-NMR data see Tables 1 and 2.

All of the known quassinoids isolated in this experiment were identified from their IR, UV, MS, and NMR (¹H, ¹³C, DEPT, ¹H⁻¹H COSY, and ¹³C⁻¹H COSY) spectra.

Acknowledgment. The authors thank Drs. M. Sugiura, K. Saiki, and T. Sai, Kobe Pharmaceutical University, for their measurements of NMR (¹H, ¹³C, DEPT, ¹H-¹H and ¹³C-¹H COSY, HMBC, and NOESY) and MS (EI and HREI) spectra and Dr. S. Ohta, Instrument Center for Chemical Analysis at Hiroshima University, for his measurement of NMR (¹H, ¹²C, DEPT, ¹H-¹H and ¹³C-¹H COSY, and NOESY) spectra. This investigation was supported in part by Grant No. CA-17625 from the National Cancer Institute awarded to K.H.L.

References and Notes

- Okano, M.; Fukamiya, N.; Lee, K. H. In *Studies in Natural Products Chemistry*; Atta-ur-Rahman, Ed.; Elsevier Science Publishers: Amsterdam, 1990; Vol. 7, pp 369-404.
- (2) Okano, M.; Lee, K. H.; Hall, I. H.; Boettner, F. E. J. Nat. Prod. 1981, 44, 470–474.
- (3) Fukamiya, N.; Okano, M.; Tagahara, K.; Aratani, T.; Muramoto, Y.; Lee, K. H. J. Nat. Prod. **1987**, 50, 1075–1079.
- (4) Okano, M.; Fukamiya, N.; Aratani, T.; Ju-Ichi, M.; Lee, K. H. J. Nat. Prod. 1985, 48, 972–975.
- (5) Fukamiya, N.; Okano, M.; Tagahara, K.; Aratani, T.; Lee, K. H. J. Nat. Prod. **1988**, *51*, 349–352.
 (6) Imamura, K.; Fukamiya, N.; Okano, M.; Tagahara, K.; Lee, K.
- (6) Imamura, K.; Fukamiya, N.; Okano, M.; Tagahara, K.; Lee, K. H. J. Nat. Prod. 1993, 56, 2091–2079.
- (7) Okano, M.; Fukamiya, N.; Toyota, T.; Tagahara, K.; Lee, K. H. J. Nat. Prod. 1989, 52, 398–401.

- (8) Toyota, T.; Fukamiya, N.; Okano, M.; Tagahara, K.; Chang, J. J.; Lee, K. H. J. Nat. Prod. 1990, 53, 1526–1532.
- (9) Okano, M.; Fujita, T.; Fukamiya, N.; Aratani, T. Bull. Chem. Soc. Jpn. 1985, 58, 1973–1800.
- (10) Matsuzaki, T.; Fukamiya, N.; Okano, M.; Fujita, T.; Tagahara, T.; Lee, K. H. J. Nat. Prod. 1991, 54, 844–848.
- (11) Daido, M.; Fukamiya, N.; Okano, M.; Tagahara, T. J. Nat. Prod. 1992, 55, 1643–1647.
- (12) Daido, M.; Fukamiya, N.; Okano, M.; Tagahara, K. J. Nat. Prod. 1995, 58, 605–608.
- (13) Lee, K. H.; Imamura, Y.; Sumida, Y.; Wu, R. Y.; Hall, I. H.; Huang, H. C. J. Org. Chem. 1979, 44, 2180–2185.
- (14) Fukamiya, N.; Okano, M.; Miyamoto, M.; Tagahara, K.; Lee, K. H. *J. Nat. Prod.* **1992**, *55*, 468–475.
 (15) Ishibashi, M.; Tsuyuki, T.; Murae, T.; Hirota, H.; Takahashi,
- (15) Ishibashi, M.; Tsuyuki, T.; Murae, T.; Hirota, H.; Takahashi, T.; Itai, A.; Iitaka, Y. *Bull. Chem. Soc. Jpn.* **1983**, *56*, 3683– 3693.
- (16) Ishibashi, M.; Murae, T.; Hirota, H.; Naora, H.; Tsuyuki, T.; Takahashi, T.; Itai, A.; Iitaka, Y. *Chem. Lett.* **1981**, 1597–1598.
- (17) Casinovi, C. G.; Ceccherelli, P.; Granolini, G.; Bellavita, V. Tetrahedron Lett. 1964, 3991–3997.
- (18) Stocklin, W.; Stefanovic, M.; Geissman, T. A.; Casinovi, C. G. *Tetrahedron Lett.* **1970**, 2399–2402.
- (19) Naora, H.; Ishibashi, M.; Furuno, T.; Tsuyuki, T.; Murae, T.; Hirota, H.; Takahashi, T.; Itai, A.; Iitaka, Y. *Bull. Chem. Soc. Jpn.* **1983**, *56*, 3694–3698.
- (20) Casinovi, C. G.; Bellavita, V.; Grandolini, G.; Ceccherelli, P. Tetrahedron Lett. 1965, 2273–2279.
- (21) Polonsky, J.; Bourguinon-Zylber, N. Bull. Soc. Chim. Fr. 1965, 2793–2799.
- (22) Ishibashi, M.; Yoshimura, S.; Tsuyuki, T.; Takahashi, T.; Itai, A.; Iitaka, Y. Bull. Chem. Soc. Jpn. 1984, 75, 2885–2892.
- (23) Polonsky, J.; Varon, Z.; Jacquemin, H.; Pettit, G. R. *Experientia* 1978, 34, 1122–1123.
- (24) Moron, J.; Rondest, J.; Polonsky, J. *Experientia* **1966**, *22*, 511–512.
- (25) M. Kupchan, J. S. M.; Lacadie, J. A. J. Org. Chem. 1975, 40, 654–656.
- (26) Furuno, T.; Naora, H.; Murae, T.; Hirota, H.; Tsuyuki, T.; Takahashi, T.; Itai, A.; Iitaka, Y.; Matsushita, K. *Chem. Lett.* **1981**, 1797–1798.
- (27) Furuno, T.; Ishibashi, M.; Naora, H.; Murae, T.; Hirota, H.; Tsuyuki, T.; Takahashi, T.; Itai, A.; Iitaka, Y. *Bull. Chem. Soc. Jpn.* **1984**, *57*, 2484–2489.
- (28) Ishibashi, M.; Murae, T.; Hirota, H.; Tsuyuki, T.; Takahashi, T.; Itai, A.; Iitaka, Y. *Tetrahedron Lett.* **1982**, *23*, 1205–1206.

NP960427C